constitutes

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HIS BYCE BROAK (AZILIO)

trade-

others. 5 lee no reason to disagree with the statement of the majority that the mark" sought to be registered may be characterized as background display. However, it makes little difference whether the marking on the label be called background display or dress of the goods, or, as the examiner described it, a mere "color scheme used by applicant as a carrying device to display the wording." Whichever it is, it is not a trademark. For a design or, for that

distinguish his own goods. alternating stripes was mere dress and not registrable, although it gave the package "a distinctive external appeardistinctive appellant's goods; that it is such distinctive appearance which is recognized by some of the purchasing public as indicating appellant's goods." This statement is directly applicable to the situation presented by the present case, it being noted that in each case evidence (consisting in the one case of affidavits and in the other of an ex parts survey) was offered to show that the applicant's design had acquired a secondary trademark, it must not only have been adopted with the intention of making it an indication of origin, but it must be recognized by the purchasing public as In In re Burgess Battery Co., 27 CCPA 1297, 112 F.2d 820, 46 USPQ 39, this court held that a design for a label consisting merely of black and white matter, for any device or mark to be a

mark by the public is unnecessary. The case of Burgess Battery Co. v. Marzall, 101 F.Supp 812, 92 USPQ, 99, 91, was decided after the adoption of the Lanham Act, a fact which the court pointed out, saying, "Because the 1946 Act did not give a new meaning to a trademark, the design cannot be considered a trademark now " \* "." supra, is still law for this court and that it was not overruled by In re Swift & Co., 42 CCPA 1048, 223 Fed 950, 106 USPQ 286, (a case which, it is sometimes argued, overruled it). In the Swift case, the court went to some pains to distinguish the case before it from the earlier case, saying, 106 USPQ at 288, "The Burgess Battery case, suprasstands for the proposition that that

think that the Burgess Battery case,

meaning. 4

The trademark now before the court origin-authenticating purpose'

s Whether, in view of the fact that the court recognized the Burgess design as being distinctive, the cases are really distinguishable on this basis, is not easy to say, but, at any rate, it is plain from reading the Swift opinion that the court did not intend to overrule the Burgess Battery case and did not think that it was 4 The board held that the poll was "not persuasive that the purchasing public in general associates the habel design sought to be registered with applicant as an identifying and distinguishing symbol," a conclusion with which I fully agree. I think it should be recognized that a survey or poll conducted ex parte by an interested party, with no possibility of checking by means of cross examination or other-

can see a possible distinction between those cases and the Swift case in that the label in the last named, with its polka dot pattern, in a decidedly more distinctive design than either of the distinction between the Burges case and the present one, but

chasing public as a trademark, it is obviously not the intention of the Lanham Act to eliminate such fundamental requirement for trademark significance. The statutory definition, in requiring that the mark be adopted and used by a manufacturer or merchant to identify and to distinguish his goods from others, most certainly was referring to their identification and distinction in the I do not think that the enactment of the Lanham Act after the Burgess Battery case, but before the Swift case, had anything to do with the latter decision. Although the Lanham Act does not spell out in its definition of the term "trademark", found in Section 45, a requirement that the symbol or device must be such as will be recognized by the purchasing public as a trademark, it is chasing public as a trademark, it is marketplace by the purchaser, and this could occur only if the purchasing public accepted the mark as an authentication. Certainly the manufacturer does not adopt a mark for the purpose of enabling, himself only to identify and

The Lanham Act never intended such a drastic and sweeping change in the law of trademarks as would be caused by the view that acceptance as a trademark by the public is unnecessary. The

consists merely of bands of color running around the can with two empty ovals superimposed upon them, the latter being obviously intended as frames for the real trademark, the autograph of the applicant, which in practice always appears upon them, but for which registration is not being asked. I think that the applicant's color scheme "does not have, as its primary function, stands for the proposition that that which is only the attractive dress of an article, although it be distinctive in its appearance and sometimes recognized by purchasers as an indication of origin, does not have, as its primary function, an origin-authenticating purpose, and is hence not a trademark entitled to federal registration under the statute." I

supra,

Particular patents-Resins

3,047,543, Morton, Littlefield, and Mecum, Sulphur - Containing Heat-Re-sistant Resins, claim 11 of application (Swift), that its "office \* \* \* is not to point out distinctly the origin or ownership of the articles to which the label is affixed" (Burgess), and that the purchasing public would not accept it as such. Hence, it is not a trademark. I would affirm the board's decision.

Application for patent of Avery A. Morton, John B. Littlefield, and William D. Mecum, Serial No. 717,307, filed Feb. 25, 1958. From decision rejecting claim 11, applicants appeal (Appeal No. 402-39). Reversed.

Before DUNCOMBE and MAGIL, Examinfor applicants.

BAILEY, Acting Examiner in

ัฑ

tions; and wherein not more than about ninety percent of the thiazole carbon attachments are in any one of

thereby characterizing the series of said attachments within the resin chain as inclusive of at least two different of the identified ortho, meta ortho, meta and para positions and para positions.

¤

No references are relied upon. [1] Claim 11 has been rejected as failing to comply with 36 U.S.C. 112. The examiner regards the claim indefinite and speculative, because, in his opinion the compound is alleged to have a complicated structure, which should have been corroborated as by giving the results of infrared spectrum analysis of

the product. We are of the opinion that this rejec-

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Patent Office Board of Appeals

Ex parte Morton, Littlefield, And Mecum

Patent issued July 31, 1962 Opinion dated Nov. 28, 1961

PATENTS

1. Claims — Indefinite — Chemical (§ 20.553)

dis-Sufficiency of Specification \_\_ S closure (§ 62.7)

claimed compound as a whole is based upon speculation; formula set out in claim cannot be held to be based on speculation in absence of some evidence that formula may be in error: mere Fact that specification shows some uncertainty as to whether Y group in formula of claim is —H or —SH is not sufficient to indicate that structure of

absence of infrared spectra analysis does not constitute such evidence; claim is not rejected as indefinite and speculative.

Appeal from Division 60. allowed.

ALLACE & CANNON and WALLACE, KINZER & DORN, both of Chicago, Ill., WALLACE

ers in Chief, and J. S. BAILEY, Acting Examiner in Chief.

This is an appeal from the final rejection of claim 11. Claims 1, 2, 3, 6 and 12, all the claims remaining in the application, have been allowed. Chief.

11. A resin having the formula The claim reads as follows:

n is at least 2 and wherein:

X is selected from —C= and =N—
o, m and p respectively identify
ortho, meta and para positions on the
ring nuclei

Y is selected from —H and —SH and wherein each of the No. 2 thizzole carbon atoms in the thiazole groups linking the thiazole groups to the ring nuclei is attached to the adjacent ring nucleus at one of the identified ortho, meta and para posi-

and Willey

v. Blackstone, Howell,

Orthuber and Stanley

.34 USPQ

Particular patents-Journal Boxes

2,856,539, Orthuber and Stanley, Apparatus and Method for Detecting Overheated Journal Boxes, refused priority as to claim 19 against Blacksstone, Howell, Willey, and Derganc application.

Patent interference No. 90,150 between Richard K. Orthuber and Charles V. Stanley, Patent No. 2,856,539, issued Oct. 14, 1958, on application filed Jan. 25, 1954, and Henry Blackstone, Sabert N. Howell, Frank G. Willey, and William Derganc, application, Serial No. 675,948, filed Aug. 2, 1957. Priority awarded to Blackstone, Howell, Willey, and Derganc; Boys, Examiner of Interferences, specially concurring with opin-10n.

C. CORNELL REMSEN, JR., RAYSON P. MORRIS, and MICHAEL EBERT, all of New York, N.Y., for Orthuber and Stanley.

ROY C. HOFGOD and JOHN M. CALL-MAPDE, both of New York, N.Y., for Blackstone, Howell, Willey, and Der-

ganc.

Before WILLNER, LEVIN, and BOYS, Examiners of Interferences.

This interference relates to detecting overheated journal boxes of railway car WILLNER, Examiner of Interferences.

axles.
The issue of the interference consists of a single count, corresponding to claim 19 of the patent to Orthuber et al., and which is as follows:

peratures a bove a predetermined value of journal boxes comprising the steps of applying a radiation image of a portion only of a moving journal box to a detector, and utilizing the signal generated by said detector for indicating that the temperature of the journal box is above said predetermined value. The method of determining

> ORTHUBER AND STANLEY V. BLACKSTONE, HOWELL, WILLEY, AND DERGANC

Opinion dated Mar. 19, 1962

Patent Office Board of Patent

Interferences

The party Blackstone et al. moved to shift the burden of proof on the basis of an earlier application, Serial No. 349,826 filed April 20, 1953. The motion to shift was granted. The now junior party Orthuber et al. moved to dissolve on the ground that Blackstone et al. has no right to make the count. This motion or pittle to make the count. This motion was denied by the Primary Examiner. The innior narty Orthuber et al. has was demo-

onstruction of specification and claims Interference counts — In general (\$ 22.501)

1. Construction PATENTS

Interference count may not have a limitation read into it by interpretation. 2. Construction of specification and

filed briefs for and appeared at final

ing the signal" clause of the count, that is, that the count requires sufficient time integration of defector excitation to produce a utilizable signal, the count reads literally upon both of the Blackstone at al. application disclosures because, as the examiner stated, only that portion of the journal box, which is "viewable" by a simple single axis optical system is applied to the detector by Blackstone et al. The interpretation put upon the count by Orthuber et al. is as if the count by Orthuber et al. is as if the count by Orthuber et al. is an ortion only of the radiation image of a moving journal box to a detector, and if it could be read thus we do not believe it would properly read upon the Blackstone disclosure. However to alter the count in this fashion is to violate the rule that a count may not have a limitation read into it by interpretation. In re Levy, 1925 C.D. 180. See also Humiston v. Voorhees, 1928 C.D. 83.

[2] We do not know if the language of the count was deliberately chosen to achieve breadth or not, but it does not introduce an ambiguity nor call back to an explicit definition of words in the specification, so it must be held to mean what it plainly says. The count being supportable by the disclosures of the senior party. Blackstone et al. the junior party is not entitled to prevail.

Accordingly, priority of invention of the subject matter in issue is hereby awarded to Henry Blackstone, Sabert N. Howell, Frank G. Willey, and William Derganc, the senior party.

Boys, Examiner of Interferences (specially concurring).

I agree with the majority in the conclusion that both of the Blackstone applications support the count in issue but for a different reason. I do not accept as sound the reasoning of the Primary Examiner, as does in effect the opinion of the majority, that since the optical system of Blackstone et al. can scan only the front and a portion of the both of the count "applying a radiation image of a portion only of a moving journal box." This view disregards the obvious impossibility of providing means to sense the radiation image of the entire exterior surface of a journal box (including back and top). Construed in this manner the count would not exclude any optically practical radiation are nowing journal boxes.

In my opinion the limitation finds ade-

cation. This portion of the specification suggests that the effective area of the sensing means be narrow in the direction of train movement but extending vertically entirely across the journabox. The expressed purpose of this arrangement is to at least approach equalization of the signal whether the train is moving fast or slowly. The sentence beginning at line 27 of the aforesaid page 7 states:

If the train is moving slowly, then the detector cell 25 may saturate before the axis box has proceeded substantially through the field of view and the output may appear as at cfor the case of DC-polarization, and at d, for AC-polarization, the will be seen that by making the detector area of elongated narrow shape, as suggested in Fig. 7 at least one is assured of an output magnitude for slow-moving trains approaching than (sic) attainable with the fast-moving trains.

saturation in the quoted excerpt indicates that the maximum signal amplitude is reached before the entire box has passed through the field of view of the radiation sensing means and that this signal is used for indicating that the temperature of the journal box is above the predetermined value as required by the count. Substantially the same language appears in the involved application of Blackstone et al. and in my opinion this provides support for the count in issue. It should be noted that the Orthuber et al. patent states in effect that their device would not satisfy the limitation in question at train speeds over fifty miles per hour for box faces of the stated dimensions. It thus appears that so far as satisfying the broad requirement of the count is concerned the only difference between the two disclosures is possibly one of the maximum speed at which the countiscent that speed could obviously be varied within wide limits by choice of parameters it is my opinion that the Blackston applications as by the Orthuber et al patent. See Heggi v. Albers-Schoenberg v. Geldermans et al., 47 CCPA 1135 believe it clear that the reference

In my opinion the above provides a sounder basis for the award of priority to Blackstone et al. in which I concur.

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industrial enamels; also printing inks and textile finishes.

mirbane oil. See nitrobenzene.

"Mirrex." 182 TM for a calendered, unplasticized PVC film. Available in film or sheeting for a wide range of packaging applications.

mirror-image molecules. See optical isomerism, enantiomorph, chiral.

misch metal. The primary commercial form of mixed rare-earth metals (95%) prepared by the electrolysis of fused rare earth chloride mixture, d approximately 6.67, mp approximately 648C. Form: waffle-like plates weighing 40-60 lb packed in oiled paper, immersed in oil, or coated with vinyl paint.

Hazard: Flammable, dangerous fire risk. Use: Lighter flints, ferrous and non-ferrous alloys, cast iron, aluminum, nickel, magnesium and copper alloys, getter in vacuum tubes, magnetic al-

miscibility. The ability of a liquid or gas to dissolve uniformly in another liquid or gas. Gases mix with one another in all proportions. This may or may not be true of liquids, whose miscibility properties depend on their chemical nature. Alcohol and water are completely miscible because of their chemical similarity, but some liquids are only partially miscible in others because of their chemical difference, e.g., benzene and water. Many gases are miscible with liquids to a greater or lesser extent, e.g., formaldehyde mixes readily with water; CO2 is partially miscible with water and oxygen only very slightly. Liquids that do not mix at all are said to be immiscible, as oil and water. The term "solubility" is often used synonymously with "miscibility" in reference to liquids, but it more properly applies to solids.

Mitchell, Peter. (1920-) A British biochemist who was the recipient of the Nobel prize for chemistry in 1978 for his work on studies of cellular energy transfer. A graduate of Cambridge and recipient of many awards, he has been Director of Research, Glynn Research Institute, since 1964.

miticide. A pesticide which kills mites, small animals of the spider class, among them the European red mite and the common red spider which infest fruit trees.

mitochondria. Particles of cytoplasm found in most respiring cells. They synthesize most of the cell's adenosine triphosphate and are the chief energy sources of living cells. They are highly plastic, mobile structures which may fragment or fuse together at random. Many enzymes, especially those involved in converting food-derived energy into a form usable by the cell, are located in the mitochondria and DNA molecules have also been found there. Yeast is a particularly rich source of mitochondria for research purposes.

mitomycin C. C<sub>15</sub>H<sub>18</sub>N<sub>4</sub>O<sub>5</sub>. Antibiotic derived from Streptomyces, stated to be effective against tumors.

mitosis. The division of a cell nucleus to produce two new cells, each having the same chemical and genetic constitution as the parent cell. The deoxyribose (nucleic acid) component of the chromosomes is present in duplicate in the original nucleus. The amount of nucleic acid is doubled just before cell division begins; subsequent events (called phases) permit separation of the products of replication to form the new nuclei. Each half-chromosome carries the identical nucleic acids of the original chromosome. See also cell (1).

Mitsunobu reaction. Intermolecular dehydration reaction occurring between alcohols and acidic components on treatment with diethyl azodicarboxylate and triphenyl phosphine under mild neutral conditions. The reaction exhibits streospecificity and regional and functional selectivity.

mixed acid. (nitrating acid). A mixture of sulfuric and nitric acids used for nitrating, e.g., in the manufacture of explosives, plastics, etc. Consists of 36% nitric acid and 61% sulfuric acid. Hazard: Spillage may cause fire or liberate dangerous gas. Causes severe burns, irritant by ingestion and inhalation, may cause NO<sub>x</sub> poisoning.

mixing. Effecting a uniform dispersion of liquid, semi-solid or solid ingredients of a mixture by means of mechanical agitation. Low-viscosity liquids and suspensions are mixed with impellers of the turbine or propellor type. The mixing action results both from direct contact of the impeller blades with the liquid and from the turbulence induced by the impeller in the outer portions of the liquid. For this reason the diameter of the impeller need be only from one-fourth to one-half that of the container. For liquids of medium viscosity, revolving paddles of various shapes are used. Thicker mixtures involving volatile solvents are mixed in closed containers (churns) equipped with fin-like members

... Minerals can be and achieve purity greater ral products. The term cally comprehends the I metals (ores) fossil fuint, glass, rocks, sulfur, he study and classifica-: chemical composition hysical, such as color, : structure. This term its to describe a variety se uses are obsolescent,

of blue pigments il dust, nuisance dust ≎arbonate troleum derivative : red

ic black pigments

asphalt of naphtha spring water containing

nd in the earth (ozocepetroleum ade by blowing air or

te term refers to such ron, copper, phosphoum, fluorine, and trace

of volume equal to ap-

il red oxide of lead. . Utah, Wisconsin.

ionoisopropanolamine.

ropyldiamidophospho-

See sweetener,

es of alkyd type resins. marketed under this odifications including s, non-drying oils, nat-

finishes including aric, hammer, and other